## PREPARATION OF 7,7,8,8-TETRACYANOQUINODIMETHANE AND ITS DERIVATIVES

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The reaction of terephthaloyl chloride with cyanotrimethylsilane in the presence of pyridine gave 1,4-bis(dicyanotrimethylsiloxymethyl)benzene, which was treated with POCl<sub>3</sub>-pyridine to afford tetracyanoquinodimethane (TCNQ) along with its dichloride. Similarly, 2,5-dimethyl-TCNQ and 2-bromo-TCNQ were prepared from the corresponding terephthaloyl chlorides.

Tetracyanoquinodimethane (TCNQ) is an excellent  $\pi$ -acceptor and charge transfer complexes of TCNQ with  $\pi$ -donors such as tetrathiafulvalene (TTF) possess high electrical conductivities. 1) In previous preparations of TCNQ, 2,3) malononitrile or highly toxic cyanogen chloride has been employed for the introduction of dicyanomethylene group of TCNQ.

CIOC COCI 
$$\stackrel{i}{\longrightarrow}$$
 TMS ON  $\stackrel{R}{\longrightarrow}$  CN  $\stackrel{$ 

Scheme 1.

We report here a new approach to TCNQ from terephthaloyl chloride with the use of cyanotrimethylsilane. Terephthaloyl chloride ( $\underline{1a}$ ) was treated with excess cyanotrimethylsilane in the presence of pyridine at 110 °C to give 1,4-bis(dicyanotrimethylsiloxymethyl)benzene ( $\underline{3a}$ ), mp 57-59 °C, in 70% yield. This compound  $\underline{3a}$  could be also prepared in 62% yield by the reaction of the same reagents, cyanotrimethylsilane and pyridine, with terephthaloyl cyanide ( $\underline{2a}$ ) which was obtained from  $\underline{1a}$  and cupper(I) cyanide in good yield. Compound  $\underline{3a}$ 

was reacted with  $POCl_3$ -pyridine at room temperature for 15 min to afford TCNQ  $(\underline{4a})$  and dichloride  $(\underline{5a})^2$  in 71% and 15% yields, respectively. In this reaction, as shown in Table 1, the dichloride  $\underline{5a}$  was obtained as the major product when the amount of pyridine was reduced. But treatment of dichloride  $\underline{5a}$  with NaBH $_4$  in 2-propanol at room temperature afforded TCNQ in 90% yield.

	Substrate	Reagent		Yield/%	
No.	<u>3a</u> (mmol)	POCl <sub>3</sub> (mmol)	Pyridine (mmol)	TCNQ <u>4a</u>	Dichloride <u>5a</u>
1	0.30	10.8	15.5	71	15
2	0.30	10.8	12.4	69	16
3	0.30	10.8	9.3	34	48
4	2.60	75.6	12.4	5	70

Table 1. The formation of TCNQ 4a and dichloride  $5a^{a}$ )

In a similar manner, 2,5-dimethyl-TCNQ  $(\underline{4b})^3$ ) and 2-bromo-TCNQ  $(\underline{4c})^3$ ) were prepared from the corresponding terephthaloyl chlorides. Thus,  $(\underline{3b})$ , mp 132-133 °C, and  $(\underline{3c})$ , mp 102-103 °C, were produced in 79% and 71% yields, respectively. Under the conditions of No.1 in the Table,  $\underline{3b}$  was converted into  $\underline{4b}$ , mp 278-280 °C,  $\underline{3b}$  and  $\underline{5b}$ ), mp 158-159 °C, in 43% and 53% yields, respectively. Treatment of  $\underline{3c}$  under the conditions of No.3 in the Table gave  $\underline{4c}$ , mp 204-206 °C,  $\underline{3b}$  in 38% yield, but  $\underline{5c}$ ) could not be isolated.

The present method may provide the most facile laboratory preparation of TCNQ and its derivatives and be widely applicable to other related systems.

References

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a) The products were isolated by silica gel column chromatography.